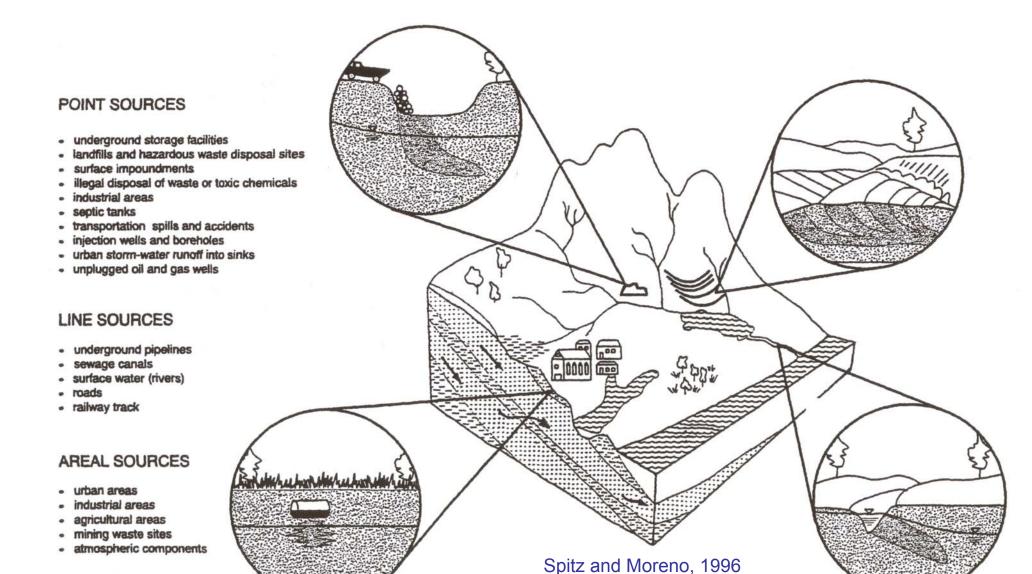
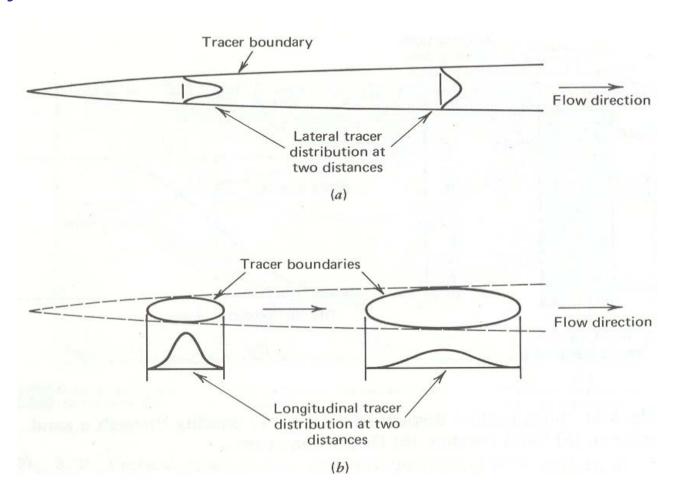
Contamination sources



Why does contaminants spread

Point injection of a tracer



Spreading processes

- Advection
- Dispersion
- Diffusion
- Sorption
- Chemical reactions
- Diffusion between mobile and immobile water

Why does contaminants spread

CHLORIDE advection and dispersion





CARBON TETRACHLORIDE advection, dispersion, and sorption





16 days

380 days

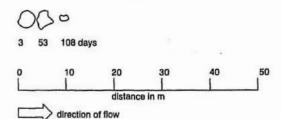
633 days

TETRACHLOROETHYLENE advection, dispersion, and sorption



16 days 380 days 633 days

TOLUENE advection, dispersion, sorption, and biodegradation



Advection

$$\mathbf{v_a} = \frac{q}{n_e} = \frac{q}{\theta_{eff}} = -\frac{K}{\theta_{eff}} \frac{dh}{dl}$$

 v_a = pore water velocity [L/T]

q = Darcy flux [L/T]

K = hydraulic conductivity [L/T]

 n_e , θ_{eff} = effective porosity [-]

dh/dl = hydraulic gradient [L/L]

Advective flux

$$q_a = qC = v_a C \theta_{eff}$$

C = solute concentration [M³/L³]

Diffusion flux

Flux of solutes from a zone with higher concentrations to a zone of lower concentrations → Ficks first law

$$\mathbf{q}_0 = -D_0 \, \frac{\partial c}{\partial x}$$

 D_0 = diffusion coefficient L²T⁻¹

C = Solute concentration ML⁻³

 q_0 = diffusive flux MT⁻¹

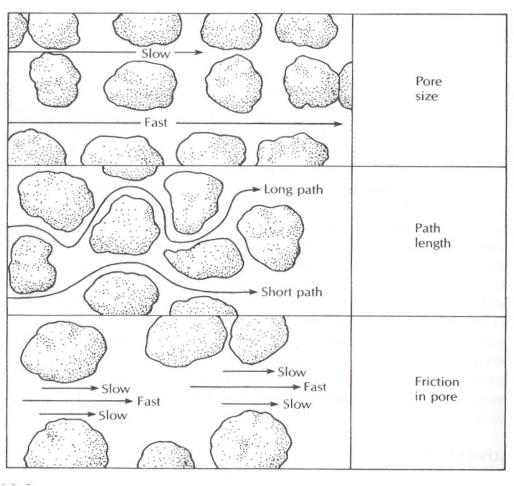
Mechanical dispersion

- Spreading due to pore to pore variations of the velocity field
- Spreading due to variations in the velocity field within the pores.
- Spreading due to incomplete knowledge regarding geological heterogeneities, source strength, source location, locale flow pattern, etc.

General assumption
$$q_M = -D_M \frac{dC}{dx}$$

Variations in the flow field – point scale

Longitudinal dispersion



Longitudinal dispersion (1D flow)

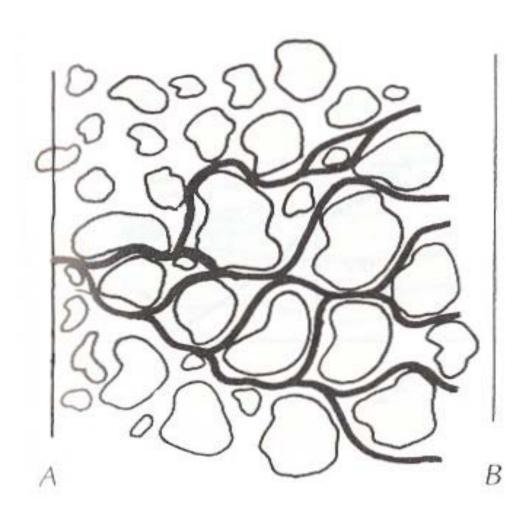
$$D_{M,L} = \alpha_L v_a$$

 a_L = Longitudinal dispersivity [L]

 v_a = Pore water velocity [L/T]

Variations in the flow field – point scale

Lateral dispersion



Lateral dispersion coefficient (1D flow)

$$D_{M,T} = \alpha_T v_a$$

 a_T = Lateral dispersivity

v_a = Pore water velocity

"Dispersion like" processes

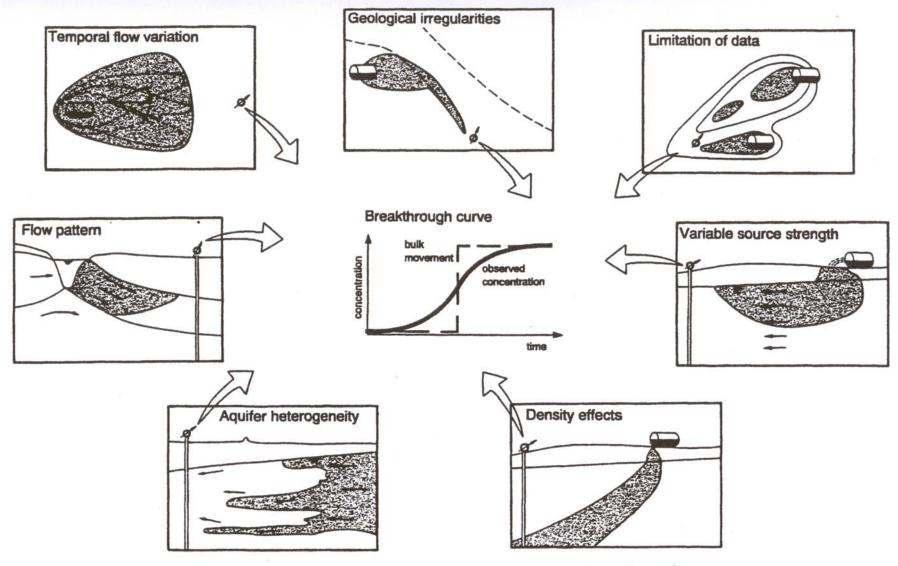
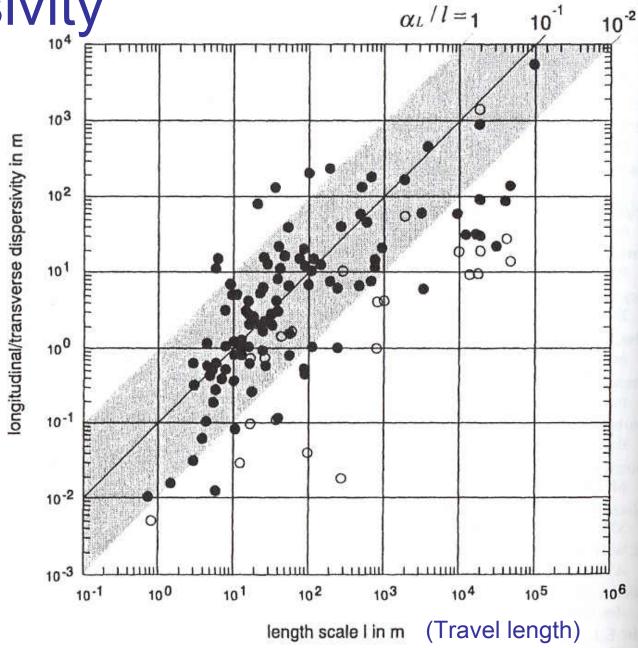


Figure 3.6 Factors which are likely to be interpreted as dispersion.

The dispersion processes are introduced because of incomplete description of advective flow ...

- scale dependent
- → "knowledge" dependent

Field dispersivity



Dispersivity

```
a_{i,j,k,l} = 4.order tensor
Is often reduces to
      = longitudinal dispersivity
a_{l}
      = lateral dispersivity in the horizontal plane
      = lateral dispersivity in the vertical plane
or simply
      = longitudinal dispersivity
a_{l}
     = lateral dispersivity
a_T
```

Dispersion coefficients

Hydrodynamic dispersion coefficients (3D flow)

$$D_{xx} = \alpha_L \frac{v_{a,x}^2}{|v_a|} + \alpha_{TH} \frac{v_{a,y}^2}{|v_a|} + \alpha_{TV} \frac{v_{a,z}^2}{|v_a|} + D^*$$

$$D_{yy} = \alpha_L \frac{v_{a,y}^2}{|v_a|} + \alpha_{TH} \frac{v_{a,x}^2}{|v_a|} + \alpha_{TV} \frac{v_{a,z}^2}{|v_a|} + D^*$$

$$D_{zz} = \alpha_L \frac{v_{a,z}^2}{|v_a|} + \alpha_{TV} \frac{v_{a,x}^2}{|v_a|} + \alpha_{TV} \frac{v_{a,y}^2}{|v_a|} + D^*$$

$$D_{xy} = D_{yx} = (\alpha_L - \alpha_{TH}) \frac{v_{a,x} v_{a,y}}{|v_a|}$$

$$D_{xz} = D_{zx} = (\alpha_L - \alpha_{TV}) \frac{v_{a,x} v_{a,z}}{|v_a|}$$

$$D_{yz} = D_{zy} = (\alpha_L - \alpha_{TV}) \frac{v_{a,y} v_{a,z}}{|v_a|}$$

 D_{xx} , D_{yy} , D_{zz} = principle components in the dispersion tensor, L²T⁻¹

 D_{xy} , D_{xz} , D_{yx} , D_{yz} , D_{zx} , D_{zy} = Cross components in the dispersion tensor, L²T⁻¹

 a_L = longitudinal dispersivity, L

 a_{TH} = horizontal lateral dispersivity, L

 a_{TV} = vertical lateral dispersivity, L

 D^* = effective molecular diffusion coefficient, L²T⁻¹

 $V_{a,x}$, $V_{a,y}$, $V_{a,z}$ = components in the pore water velocity vector, LT⁻¹

 $|v_a|$ = length of the velocity vector, LT⁻¹

Exercise

Spreading due to dispersion in a uniform stationary flow

The concentration plume that originates form a point injection follows a normal distribution with

$$\sigma_{\rm L} = \sqrt{2D_L t}$$

$$\sigma_{\mathrm{T}} = \sqrt{2D_{\mathrm{T}}t}$$

A monitoring program is set-up 1000 m down stream

Calculate s_L and s_T at the monitoring station 1000 m down stream given:

$$\alpha_{\rm L} = 10 \, \rm m$$

$$\alpha_{\rm T} = 1 \, \rm m$$

$$v_{\rm a} = 0.1 \, {\rm m/day}$$

What is s_L and s_T at the monitoring station if

$$v_a = 0.5 \,\mathrm{m/day}$$

Dispersion flux (1D flow)

Fick's law

$$q_{D} = -D_{L} \frac{dC}{dx}$$

 D_1 = Longitudinal dispersion coefficient L²T⁻¹

C = Solute concentration ML⁻³

Dispersion and diffusion flux (1D strømning)

$$q_h = -D_h \frac{dC}{dx}$$

C = Solute concentration ML⁻³

 D_h = Longitudinal hydrodynamic dispersion coefficient L²T⁻¹

$$D_h = D_L + D_0 = \alpha_L v_a + D_0$$

Total flux

Advection + dispersion (1D flow)

$$q_{tot} = qC - D_h \frac{dC}{dx}$$

 $q_{tot} = Total flux MT^{-1}L^{-2}$

q = darcy velocity LT⁻¹

Total flux

advection + dispersion (3D flow)

$$q_{\text{tot,x}} = q_x C - D_{xx} \frac{dC}{dx} - D_{xy} \frac{dC}{dy} - D_{xz} \frac{dC}{dz}$$

Governing equation

1D advection-dispersion equation

$$\frac{\partial \theta C}{\partial t} = D_L \frac{\partial^2 \theta C}{\partial x^2} - \frac{\partial \theta v_a C}{\partial x} - S$$

 D_L = longitudinal hydrodynamic dispersion coefficient, L²T⁻¹

C = solute concentration, ML^{-3}

 v_a = Pore water (transport) velocity LT⁻¹

S = sink/source due to sorption or decay, ML-3T-1

t = time, T

Governing equation

3D advection-dispersion equation

$$\frac{\partial \theta C}{\partial t} = \frac{\partial}{\partial x_i} \left(\theta D_{ij} \frac{\partial C}{\partial x_j} \right) - \frac{\partial \theta v_{a,i} C}{\partial x_i} - S$$

 D_{l} = components in the hydrodynamic dispersion tensor, L²T⁻¹

C = solute concentration, ML^{-3}

 v_a = components in the pore water velocity vector LT⁻¹

S = sink/source due to sorption or decay, ML-3T-1

 R_n = chemical reactions, ML⁻³T⁻¹

t = time, T

Sorption as a sink/source term, S_i

$$S_i = -\rho_b \, \frac{\partial C_a}{\partial t}$$

 $r_b = bulk density, ML^{-3}$

 C_a = adsorbed concentration, MM⁻¹

$$\frac{\partial \theta C}{\partial t} = \frac{\partial}{\partial x_i} \left(\theta D_{ij} \frac{\partial C}{\partial x_j} \right) - \frac{\partial \theta v_{a,i} C}{\partial x_i} - \rho_b \frac{\partial C_a}{\partial t}$$

$$\downarrow$$

$$\frac{\partial \theta C}{\partial t} + \rho_b \frac{\partial C_a}{\partial t} = \frac{\partial}{\partial x_i} \left(\theta D_{ij} \frac{\partial C}{\partial x_j} \right) - \frac{\partial \theta v_{a,i} C}{\partial x_i}$$

$$\downarrow$$

$$R \frac{\partial \theta C}{\partial t} = \frac{\partial}{\partial x_i} \left(\theta D_{ij} \frac{\partial C}{\partial x_j} \right) - \frac{\partial \theta v_{a,i} C}{\partial x_i}$$

where
$$R = 1 + \frac{\rho_b}{\theta} \frac{\partial C_a}{\partial C}$$
 (Retardation factor)

Sorption

Equilibrium controlled sorption

Linear sorption

$$R = 1 + \frac{\rho_b}{\theta} \frac{\partial C_a}{\partial C} = 1 + \frac{\rho_b}{\theta} K_d$$

where

 K_d = distribution coefficient [L³M⁻¹] $C_a = K_d$ C

$$R\theta \frac{\partial C}{\partial t} = R_{i,j,k}\theta_{i,j,k} \frac{C_{i,j,k}^{n+1} - C_{i,j,k}^{n}}{\Delta t}$$

Sorption

Non-equilibrium controlled sorption

$$\frac{\partial \theta C}{\partial t} + \rho_b \frac{\partial C_a}{\partial t} = \frac{\partial}{\partial x_i} \left(\theta D_{ij} \frac{\partial C}{\partial x_j} \right) - \frac{\partial \theta v_{a,i} C}{\partial x_i}$$

$$\rho_b \frac{\partial C_a}{\partial t} = \beta \left(C - \frac{C_a}{K_d} \right)$$

b = First order mass transfer rate between dissolved and sorbed phases, T⁻¹

K_d ≠ distribution coefficient for the sorbed phase

$$\rho_{b(i,j,k)} \frac{C_{a,i,j,k}^{n+1} - C_{a,i,j,k}^{n}}{\Delta t} = \beta_{i,j,k} \left(C_{i,j,k}^{n+1} - \frac{C_{a,i,j,k}^{n+1}}{K_{d(i,j,k)}} \right)$$

$$S_{i} = \beta_{i,j,k} \left(C_{i,j,k}^{n+1} - \frac{C_{a,i,j,k}^{n+1}}{K_{d(i,j,k)}} \right)$$

Decay/degradation

first order

$$S_i = (\lambda_1 \theta C + \lambda_2 \rho_b C_a)$$
 Decay of dissolved solute
Decay of sorbed solute

 I_1 = first order reactions rate for dissolved solute, T⁻¹

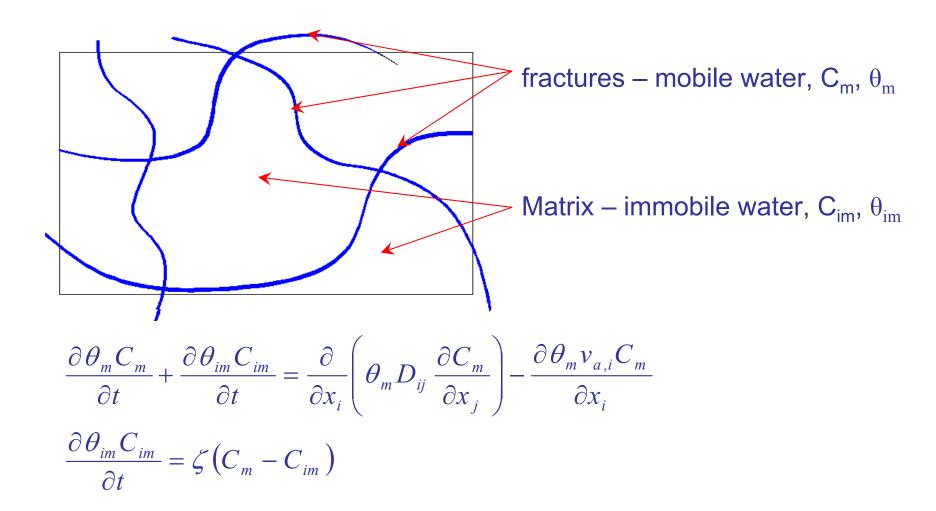
 I_2 = first order reaction rate for sorbed solute, T^{-1}

$$\lambda = \left(\ln 2\right) / t_{\frac{1}{2}}$$

 $T_{\frac{1}{2}}$ = half life, T

$$S_{i} = -\lambda_{1} \theta_{i,j,k} C_{i,j,k}^{n} - \lambda_{2} \rho_{b(i,j,k)} C_{a,i,j,k}^{n}$$

Dual porosity systems



 ξ = Mass transfer rate between mobile and immobile water, T⁻¹

Dual porosity systems

$$\frac{\partial \theta_{m} C_{m}}{\partial t} + \frac{\partial \theta_{im} C_{im}}{\partial t} = \frac{\partial}{\partial x_{i}} \left(\theta_{m} D_{ij} \frac{\partial C_{m}}{\partial x_{j}} \right) - \frac{\partial \theta_{m} v_{a,i} C_{m}}{\partial x_{i}}$$

$$\frac{\partial \theta_{im} C_{im}}{\partial t} = \mathcal{L} \left(C_{m} - C_{im} \right)$$

$$\theta_{im(i,j,k)} \frac{C_{im(i,j,k)}^{n+1} - C_{im(i,j,k)}^{n}}{\Delta t} = \zeta \left(C_{m(i,j,k)}^{n+1} - C_{im(i,j,k)}^{n+1} \right)$$

$$R_{i} = \zeta \left(C_{m(i,j,k)}^{n+1} - C_{im(i,j,k)}^{n+1} \right)$$

Solution of the 3D advection-dispersion equation

$$\frac{\partial}{\partial x_i} \left(D_{ij} \frac{\partial C}{\partial x_j} \right) - v_{a,i} \frac{\partial C}{\partial x_i} + S_n = \frac{\partial C}{\partial t}$$

- Standard finite difference methods
- Particle methods (random walk)
- Hybrid methods (MOC)
- High order FD or FV methods (TVD)